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Search Result customer reviews for algorithms design and analysis by udit agarwal pdf Free book download - free ebooks for pdf - free webThis invention relates to a process for preparation of a copper-based glass, and more specifically, to a process for preparation of a copper-based glass excellent in formability. It is well-known that, in spite of the fact that copper is poor in corrosion resistance and wear resistance, copper having excellent formability is used for a glass sheet widely employed in a window glass of an automobile, a general window glass and the like. Copper is not only superior to zinc in corrosion resistance but also superior to tin in formability. As compared with zinc or tin, copper is inferior in strength and excellent in formability. Accordingly, when glass which is excellent in formability is prepared, copper is generally employed as a raw material. However, when a glass sheet having a low coefficient of thermal expansion is prepared, the glass sheet is poor in formability. Therefore, to achieve a low thermal expansion property, it is necessary to add a suitable amount of expensive stabilizing agent, but this has caused a problem of cost increase. To solve the problem, in European Patent No. 335890, there is disclosed a glass excellent in formability having a coefficient of thermal expansion of $5 \times 10^{-7} / ^\circ\text{C}$ to $8 \times 10^{-7} / ^\circ\text{C}$ and containing, as an essential component, from 2% to 15% by weight of CuO and SiO_2 , but this glass is not satisfactory in view of cost. In order to prepare a copper-based glass excellent in formability, there has been known a process which comprises adding a tin compound to an aqueous solution of copper acetate, adjusting pH of the aqueous solution, adding an alkaline solution, forming a copper-tin hydrate, making the hydrate react with an alkali, allowing the hydrate to be substantially completely dissolved in an aqueous solution, neutralizing the alkali and heat-treating the neutralized solution (see Japanese Patent Application Laid-open No. 59-170031). This process is very effective when the aqueous solution of copper acetate is employed in an amount of 20% by weight or more, but when an aqueous solution of copper acetate containing a slight amount of copper is employed, no copper- 6d1f23a050

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